## Crystallographic Structure Determination of a Triterpene Dimethyl Ester $\epsilon$ -Lactone from Dammar Resin

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Summary A triterpenoid dimethyl ester  $\epsilon$ -lactone isolated from dammar resin has been shown by X-ray crystallographic examination to be 23-hydroxy-2,3-secours-12-ene-2,3,28-trioic acid ( $2 \rightarrow 23$ )-lactone 3,28-dimethyl ester (IV): it is probably formed biogenetically by oxidation of asiatic acid (II).

Some years ago a triterpenoid unsaturated dimethyl ester lactone ("compound B"), m.p. 234—236°,  $[\alpha]_D + 117°$  was isolated<sup>1</sup> from dammar resin in small quantity along with ursonic acid (I) and asiatic acid (II).<sup>2</sup> The elemental analysis and i.r. spectrum of "compound B" indicated a molecular formula of  $C_{32}H_{50-52}O_6$  and the presence of two methoxycarbonyl groups and a  $\delta$ -lactone. Its u.v. spectrum suggested that it might be an ursane derivative. Lack of material has prevented further chemical work on this compound but the recent developments in X-ray crystallography have now enabled the structure to be determined on one of the few remaining crystals.

"Compound B" forms triclinic crystals in the non-centrosymmetric space group P1, with Z = 1, and a = 7.01, b = 9.16, c = 11.72 Å,  $\alpha = 90.62$ ,  $\beta = 92.03$ , and  $\gamma =$ 100.42°. 3180 reflections out to a Bragg angle of 80° were collected on a computer-controlled, four-circle diffractometer using Ni-filtered Cu- $K_{\alpha}$  radiation. The structure was solved with a computer programme based on the tangent formula<sup>3</sup> applied in a manner similar to that suggested by Germain and Woolfson.<sup>4</sup> The molecular structure is the dimethyl ester  $\epsilon$ -lactone (IV) with the  $HO_2$ molecular formula  $C_{32}H_{48}O_6$ : the original analysis is correct for this formula. So far 47 of the 48 hydrogen atoms have been placed with difference-Fourier syntheses, and R is 0.05. All bond lengths and angles are as expected except for some longer bonds (1.58-1.59 Å) between highly substituted carbon atoms (e.g. C-9 and C-14). The presence



of the  $\epsilon$ -lactone group is novel in the pentacyclic triterpenefield.

The biogenetic origin of "compound B" can be explained.

by assuming that it originates from asiatic acid (II) with which it occurs in dammar resin. Oxidation of the  $2\alpha, 3\beta$ diol grouping in asiatic acid could lead to the 2,3-seco-trioic acid (III). Lactonisation of the C-2 carboxy-group with the C-23 hydroxy-group and esterification of the free carboxyl groups would then afford "compound B." A naturally occurring 2,3-seco-dioic acid (V)<sup>5</sup> is already known.

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